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Physical Organic Chemistry

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1561	Kinetics and mechanism of nitrosation of clonidine: a bridge between nitrosation of amines and ureas	(I) = (I)
	Fátima Norberto, José A. Moreira, Eduarda Rosa, Jim Iley, J. Ramón Leis and M. Elena Peña	(I) = (I)
1567	O-H Bond strengths and one-electron reduction potentials of multisubstituted phenols and phenoxyl radicals. Predictions using free energy relationships	$R^{6} \rightarrow R^{2}$ $R^{5} \rightarrow R^{3}$
	Mats Jonsson, Johan Lind, Trygve E. Eriksen and Gabor Merényi	A general equation for determination of O–H bond strengths of multisubstituted phenols based on linear free energy relationships is presented
1569	Conformational effects in β-functional acyclic organosilicon compounds	
		R ₃ SiCH ₂ CH ₂ X I
	Martin S. Beevers, Susan Bratt, Ann W. P. Jarvie and Michael C. Perry	Theoretical and experimental evaluation of R and X on the conformation of I
1575	Reactivity-selectivity relationship and kinetic solvent isotope effects in nucleophilic substitution reactions	
	Substitution (Cactions	$\log k_{\rm N}/k_{\rm S} = \Delta \rho_{\rm Y} \times \sigma_{\rm Y}$ $\log \rm KSIE = \Delta \rho_{\rm Y} \times \sigma_{\rm Y}$
	Ikchoon Lee, Han Joong Koh, Young Sook Park and Hai Whang Lee	Signs of $\Delta \rho_{\rm Y}$ are indicative of reaction mechanism

1583	The superoxide radical reacts with tyrosine- derived phenoxyl radicals by addition rather than by electron transfer	HO_2CCHNH_2 $\downarrow CO_2H$ $\downarrow CO_2H$ $HO_2CC_HNH_2$ $\downarrow CO_2H$ HO_2C_HNH H_2O_2
	Faming Jin, Johannes Leitich and Clemens von Sonntag	The superoxide radical reacts with tyrosine-derived phenoxyl radicals by addition rather than by electron transfer
1589	Preparation and characterisation of thienonaphthoquinones and their radical ions	S R^1 R^2 R^2 R^2
	Joe A. Crayston, Ahmed Iraqi, Philip Mallon and John C. Walton	Redox reactions of three naphtho[2,3-c]thiophene-4,9-diones were studied electrochemically and by EPR spectroscopy
1597	Proton transfer in rubazoic acid derivatives in solution and in the solid state. An NMR study	$ \begin{array}{c} R^{1} & R^{1} \\ N \\ R^{2} \\ N \\ O - H \end{array} \xrightarrow{N}_{Q} \\ O - H \end{array} \xrightarrow{N}_{Q} \\ R^{2} \\ R^{2} \\ R^{2} \\ N \\ O \\ H - O \end{array} \xrightarrow{N}_{Q} \\ R^{2} \\ R^{2}$
	Alejandro C. Olivieri, Dionísia Sanz, Rosa M ^a . Claramunt and José Elguero	¹ H, ¹³ C, ¹⁵ N and ² H NMR in solution, and ¹³ C solid-state NMR data are reported for rubazoic acid derivatives
1603	Photofragmentation voltammetry studies of the aminocarbene complexes $[(\eta^5-C_5H_5)Fe(CO)(L) \{=C(NHR^1)(R^2)\}]BF_4$ $[L = PPh_3, P(p-Tol)_3; R^1 = Me, CH_2Ph;$ $R^2 = Me, Bu]$	$ \begin{array}{c} $
	Stephen G. Davies, Michael R. Metzler, W. Carl Watkins, Richard G. Compton, Jonathan Booth and John C. Eklund	L = PPh ₃ , P(ρ -Toi) ₃ R ¹ = Me, CH ₂ Ph; R ² = Me, Bu
1611	Luminescence and charge transfer. Part 3. The use of chromophores with ICT (internal charge transfer) excited states in the construction of fluorescent PET (photoinduced electron transfer) pH sensors and related absorption pH sensors with aminoalkyl side chains	NO ₂ HICH _{2l2} NEt ₂ NHICH
	A. Prasanna de Silva, H. Q. Nimal Gunaratne, P. L. Mark Lynch, Alan J. Patty and Graham L. Spence	1 2 3 1-3 are absorption pH sensors while 2 and 3 also show fluorescent pH sensor action
1617	The strain limit in intramolecular nucleophilic substitution	ArSO ₂ 3 2 OMes ArSO ₂ 4rSO ₂ ArSO ₂ ArSO ₂
	Stephen M. Jeffery and Charles J. M. Stirling	Arso ₂

1625	Investigation and rationalisation of hydrogen bonding patterns in sulfonylamino compounds and related materials: crystal structure determination of microcrystalline solids from powder X-ray diffraction data Philip Lightfoot, Maryjane Tremayne, Christopher Glidewell, Kenneth D. M. Harris and Peter G. Bruce	CH_3 SO_2NH_2 SO_2NHNH_2 CH_3 SO_2NHNH_2 The crystal structures of these compounds, solved from powder X-ray diffraction data, are reported and their hydrogen bonding patterns, and those of related materials, are discussed in terms of graph set analysis
1631	Preparation, properties and structure of crystalline silver ketenide (μ-oxoethenylidene disilver) and its pyridine complex	
		$2AgOAc \xrightarrow{CH_2=C=O} Ag_2C=C=O \cdot Py \xrightarrow{Heat} Ag_2C=C=O$
	Ernest T. Blues, Derek Bryce-Smith, Ronald Shaoul, Hans Hirsch and Michael J. Simons	Silver ketenide and its pyridine complex have unusual structures involving monolayers of silver atoms separated by rod-like ketenide groups normal to the sheets of silver atoms
1643	Acid-catalysed hydrolysis and protonation behaviour of <i>N</i> -arylpropane-1,3-sultams	
		1 R = a, MeO; b, Me; c, H; d, Cl; e, NO ₂
	Yunus Bekdemir, John G. Tillett and Romuald I. Zalewski	The rate maxima observed in the acid-catalysed hydrolyses of sultams are shown to be associated with extensive protonation of these substrates in strong acid
1647	Origin of regioselectivity in the <i>O</i> -methylation of erythromycin as elucidated with the aid of computational conformational space search	HO HO HO HO HO HO HO HO HO HO HO HO HO H
	Hitoshi Gotō, Yutaka Kawashima, Masato Kashimura, Shigeo Morimoto and Eiji Ōsawa	generates low-energy conformers
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1655	Synthesis and studies on surface and self- assembly properties of polyphenylsulfonates in aqueous solution. Part 1: sodium 5'-phenyl- 1,1': 3',1"-terphenyl-4-sulfonate	
	Jan Czapkiewicz, Piotr Milart and Bożena Tutaj	SO ₃ Na
1659	Dual reaction channels for solvolyses of acyl chlorides in alcohol-water mixtures	$Bu'C_{CI}^{(0)} \xrightarrow{ROH-H_2O} Bu'C_{OH}^{(0)} + Bu'C_{OR}^{(0)}$
	T. William Bentley and Chang Sub Shim	Rates and products of the above reaction are interpreted in terms of both competing nucleophiles, and competing reaction channels in contrast to the transition state variation model



Corrigendum

- 1701 The reaction of imidazole with some 1-halogeno-2,4-dinitrobenzenes Elizabeth T. Akinyele, Delanson F. Crist and Jack Hirst
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